GOVERNMENT COLLEGE FOR WOMEN (AUTONOMOUS), SRIKAKULAM

DEPARTMENT OF PHYSICS Paper: HEAT & THERMODYNAMICS

CLASS:II MPCS, MPC & MPWT	MID 1 EXAM		Y	EAR 2	023-24				
EXAM DATES:5 TH -7 TH OCT 2023	ТІМЕ:2Н				Marks:30				
	SECTION-A	1X8=8M							
ANSWER THE FOLLOWING									
1. A] Deduce an expression for the thermal conductivity of a gas on the basis of kinetic theory of gases?									
(OR) B] Explain the change in entropy of reversible and irre	eversible processes?								
	SECTION-B	3X4=12M							
ANSWER THE THREE OF THE FOLLOWING									
2. What do you mean by transport phenomena?									
3. What is the mean by mean free path of the molecule of gas?									
4. Write the assumptions of kinetic theory of gases?									
5. A Thermos flask containing coffee is vigorously shaken.the iii]has work been done on it? & iv]has its internal energy		I]does its temperature rise?	ii] ha	s heat l	been added to it?				
6. Derive an expression for the work done in an isothermal process	s?								
7. Explain the change in entropy when ice changes into steam?									
	SECTION-C	5X2=10M							
ANSWER ALL THE FOLLOWING									
8. Define mean free path? 9. Explain heat and temperature? 10.Calculate the efficiency of reversible engine operates between the temperatures 200°c and 120°c									
11. Write the failures of Carnot's engine? 12. Write the uses of T-S diagram?									
	SECTION-C	5X1=5M							
ANSWER THE FOLLOWING									
13.K.E. of gas molecules is zero at: (a) 0°C (b) 273°C (c) -273°C (d) none of the above			[]					
14.For an ideal gas			[]					
a]Size of molecules is negligible. (b) There is no force of attraction or repulsion amongst the molecules.(c) It strictly obeys the gas laws. (d) All of the above.									
15.What is thermodynamics? a) study of the relationship between heat and other forms of energy be c) study of the relationship between mechanical energy to other form				ns of ene] ergy				
16.What is the value of the absolute thermodynamic temperature a) $3K-b)\ 0K-c)\ 1K-d)\ 4K$	e scale?			[]				
17. Which of the following is true?]]				
a) for an isolated system, dS>=0 b) for a reversible process, dS=0 c)	for an irreversible process, dS>	·0 d) all of the mentioned							